



**WIKIPEDIA**  
The Free Encyclopedia

# Energy density Extended Reference Table

---

This is an extended version of the energy density table from the main [Energy density](#) page:

Energy densities table

Storage type	Specific energy (MJ/kg)	Energy density (MJ/L)	Peak recovery efficiency %	Practical recovery efficiency %
Arbitrary Antimatter	89,875,517,874	depends on density		
Deuterium–tritium fusion	576,000,000 <sup>[1]</sup>			
Uranium-235 fissile isotope	144,000,000 <sup>[1]</sup>	1,500,000,000		
Natural uranium (99.3% U-238, 0.7% U-235) in fast breeder reactor	86,000,000			
Reactor-grade uranium (3.5% U-235) in light-water reactor	3,456,000			30%
Pu-238 α-decay	2,200,000			
Hf-178m2 isomer	1,326,000	17,649,060		
Natural uranium (0.7% U235) in light-water reactor	443,000			30%
Ta-180m isomer	41,340	689,964		
Metallic hydrogen (recombination energy)	216 <sup>[2]</sup>			
Specific orbital energy of Low Earth orbit (approximate)	33.0			
Beryllium + Oxygen	23.9 <sup>[3]</sup>			
Lithium + Fluorine	23.75			
Octaazacubane potential explosive	22.9 <sup>[4]</sup>			
Ammonia (NH <sub>3</sub> )	16.9	11.5 <sup>[5]</sup>		
Hydrogen + Oxygen	13.4 <sup>[6]</sup>			
Gasoline + Oxygen → Derived from Gasoline	13.3			
Dinitroacetylene explosive - computed	9.8			
Octanitrocubane explosive	8.5 <sup>[7]</sup>	16.9 <sup>[8]</sup>		
Tetranitrotetrahedrane explosive - computed	8.3			
Heptanitrocubane explosive - computed	8.2			
Sodium (reacted with chlorine)	7.0349			
Hexanitrobenzene explosive	7 <sup>[9]</sup>			
Tetranitrocubane explosive - computed	6.95			
Ammonal (Al+NH <sub>4</sub> NO <sub>3</sub> oxidizer)	6.9	12.7		
Storage type	Energy density by mass (MJ/kg)	Energy density by volume (MJ/L)	Peak recovery efficiency %	Practical recovery efficiency %

Storage type	Specific energy (MJ/kg)	Energy density (MJ/L)	Peak recovery efficiency %	Practical recovery efficiency %
Tetranitromethane + hydrazine bipropellant - computed	6.6			
Nitroglycerin	6.38 <sup>[10]</sup>	10.2 <sup>[11]</sup>		
ANFO-ANNM	6.26			
battery, Lithium-air	6.12			
Octogen (HMX)	5.7 <sup>[10]</sup>	10.8 <sup>[12]</sup>		
TNT [Kinney, G.F.; K.J. Graham (1985). <i>Explosive shocks in air</i> . Springer-Verlag. ISBN 978-3-540-15147-0.]	4.610	6.92		
Copper Thermite (Al + CuO as oxidizer)	4.13	20.9		
Thermite (powder Al + Fe <sub>2</sub> O <sub>3</sub> as oxidizer)	4.00	18.4		
Hydrogen peroxide decomposition (as monopropellant)	2.7	3.8		
battery, Lithium-ion nanowire	2.54			95% <sup>[13]</sup>
battery, Lithium Thionyl Chloride (LiSOCl <sub>2</sub> ) <sup>[14]</sup>	2.5			
Water 220.64 bar, 373.8 °C	1.968	0.708		
Kinetic energy penetrator	1.9	30		
battery, Fluoride-ion	1.7	2.8		
battery, Hydrogen closed cycle H fuel cell <sup>[15]</sup>	1.62			
Hydrazine decomposition (as monopropellant)	1.6	1.6		
Ammonium nitrate decomposition (as monopropellant)	1.4	2.5		
Thermal Energy Capacity of Molten Salt	1			98% <sup>[16]</sup>
Molecular spring approximate	1			
battery, Sodium-Sulfur	0.72 <sup>[17]</sup>	1.23		85% <sup>[18]</sup>
battery, Lithium-Manganese <sup>[19][20]</sup>	0.83-1.01	1.98-2.09		
battery, Lithium-ion <sup>[21][22]</sup>	0.46-0.72	0.83-3.6 <sup>[23]</sup>		95% <sup>[24]</sup>
battery, Lithium-Sulfur <sup>[25]</sup>	1.80 <sup>[26]</sup>	1.26		
battery, Sodium-Nickel Chloride, High Temperature	0.56			
battery, Silver-oxide <sup>[19]</sup>	0.47	1.8		
Storage type	Energy density by mass (MJ/kg)	Energy density by volume (MJ/L)	Peak recovery efficiency %	Practical recovery efficiency %

Storage type	Specific energy (MJ/kg)	Energy density (MJ/L)	Peak recovery efficiency %	Practical recovery efficiency %
Flywheel	0.36-0.5 <sup>[27][28]</sup>			
5.56 × 45 mm NATO bullet	0.4	3.2		
battery, Nickel–metal hydride (NiMH), low power design as used in consumer batteries <sup>[29]</sup>	0.4	1.55		
battery, Zinc-manganese (alkaline), long life design <sup>[19][21]</sup>	0.4-0.59	1.15-1.43		
Liquid Nitrogen	0.349			
Water - Enthalpy of Fusion	0.334	0.334		
battery, Zinc Bromine flow (ZnBr) <sup>[30]</sup>	0.27			
battery, Nickel metal hydride (NiMH), High Power design as used in cars <sup>[31]</sup>	0.250	0.493		
battery, Nickel–Cadmium (NiCd) <sup>[21]</sup>	0.14	1.08		80% <sup>[24]</sup>
battery, Zinc–Carbon <sup>[21]</sup>	0.13	0.331		
battery, Lead–acid <sup>[21]</sup>	0.14	0.36		
battery, Vanadium redox	0.09	0.1188		70-75%
battery, Vanadium–Bromide redox	0.18	0.252		80%–90% <sup>[32]</sup>
Capacitor Ultracapacitor	0.0199 <sup>[33]</sup>	0.050		
Capacitor Supercapacitor	0.01		80%–98.5% <sup>[34]</sup>	39%–70% <sup>[34]</sup>
Superconducting magnetic energy storage		0.008 <sup>[35]</sup>		>95%
Capacitor	0.002 <sup>[36]</sup>			
Neodymium magnet		0.003 <sup>[37]</sup>		
Ferrite magnet		0.0003 <sup>[37]</sup>		
Spring power (clock spring), torsion spring	0.0003 <sup>[38]</sup>	0.0006		
Storage type	Energy density by mass (MJ/kg)	Energy density by volume (MJ/L)	Peak recovery efficiency %	Practical recovery efficiency %

## Notes

1. Prelas, Mark (2015). *Nuclear-Pumped Lasers* ([https://books.google.com/books?id=Hmn\\_CgAAQBAJ&pg=PA135](https://books.google.com/books?id=Hmn_CgAAQBAJ&pg=PA135)). Springer. p. 135. ISBN 9783319198453.
2. [http://iopscience.iop.org/1742-6596/215/1/012194/pdf/1742-6596\\_215\\_1\\_012194.pdf](http://iopscience.iop.org/1742-6596/215/1/012194/pdf/1742-6596_215_1_012194.pdf)
3. Cosgrove, Lee A.; Snyder, Paul E. (2002-05-01). "The Heat of Formation of Beryllium Oxide". *Journal of the American Chemical Society*. 75 (13): 3102–3103. doi:10.1021/ja01109a018 (<https://doi.org/10.1021%2Fja01109a018>).

4. Glukhovtsev, Mikhail N.; Jiao, Haijun; Schleyer, Paul von Ragué (1996-05-28). "Besides N<sub>2</sub>, What Is the Most Stable Molecule Composed Only of Nitrogen Atoms?†". *Inorganic Chemistry*. 35 (24): 7124–7133. doi:10.1021/ic9606237 (<https://doi.org/10.1021%2Fic9606237>). PMID 11666896 (<https://pubmed.ncbi.nlm.nih.gov/11666896/>).
5. Ammonia#Combustion
6. Miller, Catherine (1 February 2021). "Introduction to Rocket Propulsion" ([https://www.cs.middlebury.edu/~cm2/personal-website/lecture\\_notes/lecture6.pdf](https://www.cs.middlebury.edu/~cm2/personal-website/lecture_notes/lecture6.pdf)) (PDF). Retrieved 9 May 2021.
7. Wiley Interscience (<https://archive.today/20130105062703/http://www3.interscience.wiley.com/journal/122324589/abstract>)
8. Octanitrocubane
9. Wiley Interscience (<https://archive.today/20130105055433/http://www3.interscience.wiley.com/journal/109618256/abstract>)
10. "Chemical Explosives" (<http://www.fas.org/man/dod-101/navy/docs/es310/chemistry/chemistry.htm>). Fas.org. 2008-05-30. Retrieved 2010-05-07.
11. Nitroglycerin
12. HMX
13. "Nanowire battery can hold 10 times the charge of existing lithium-ion battery" (<http://news-service.stanford.edu/news/2008/january9/nanowire-010908.html>). News-service.stanford.edu. 2007-12-18. Retrieved 2010-05-07.
14. "Lithium Thionyl Chloride Batteries" (<https://web.archive.org/web/20090204131145/http://nexergy.com/lithium-thionyl-chloride.htm>). Nexergy. Archived from the original (<http://www.nexergy.com/lithium-thionyl-chloride.htm>) on 2009-02-04. Retrieved 2010-05-07.
15. "The Unitized Regenerative Fuel Cell" (<https://web.archive.org/web/20080920152815/https://www.llnl.gov/str/Mitlit.html>). Llnl.gov. 1994-12-01. Archived from the original (<http://www.llnl.gov/str/Mitlit.html>) on 2008-09-20. Retrieved 2010-05-07.
16. "Technology" (<https://web.archive.org/web/20080119090114/http://www.solar-reserve.com/technology.html>). SolarReserve. Archived from the original (<http://www.solar-reserve.com/technology.html>) on 2008-01-19. Retrieved 2010-05-07.
17. "New battery could change world, one house at a time" ([https://web.archive.org/web/20151017163418/http://www.heraldextra.com/news/article\\_b0372fd8-3f3c-11de-ac77-001cc4c002e0.html](https://web.archive.org/web/20151017163418/http://www.heraldextra.com/news/article_b0372fd8-3f3c-11de-ac77-001cc4c002e0.html)). Heraldextra.com. 2009-04-04. Archived from the original ([http://www.heraldextra.com/news/article\\_b0372fd8-3f3c-11de-ac77-001cc4c002e0.html](http://www.heraldextra.com/news/article_b0372fd8-3f3c-11de-ac77-001cc4c002e0.html)) on 2015-10-17. Retrieved 2010-05-07.
18. Kita, A.; Misaki, H.; Nomura, E.; Okada, K. (August 1984). "Energy Citations Database (ECD) -- Document #5960185". *Proc., Intersoc. Energy Convers. Eng. Conf.; (United States)*. Osti.gov. 2. OSTI 5960185 (<https://www.osti.gov/biblio/5960185>).
19. "ProCell Lithium battery chemistry" (<https://web.archive.org/web/20110710160212/http://www1.duracell.com/procell/chemistries/lithium.asp>). Duracell. Archived from the original (<http://www.duracell.com/Procell/chemistries/lithium.asp>) on 2011-07-10. Retrieved 2009-04-21.
20. "Properties of non-rechargeable lithium batteries" (<http://www.corrosion-doctors.org/PrimBatt/table2.htm>). corrosion-doctors.org. Retrieved 2009-04-21.
21. "Battery energy storage in various battery types" (<https://web.archive.org/web/20090428060954/http://www.allaboutbatteries.com/Battery-Energy.html>). AllAboutBatteries.com. Archived from the original (<http://www.allaboutbatteries.com/Battery-Energy.html>) on 2009-04-28. Retrieved 2009-04-21.

22. A typically available lithium-ion cell with an Energy Density of 201 wh/kg "Li-Ion 18650 Cylindrical Cell 3.6V 2600mAh - Highest Energy Density Cell in Market (LC-18650H4) - LC-18650H4" ([http://web.archive.org/web/20081201065328/http://www.batteryspace.com/index.asp?PageAction=VIEWPROD&ProdID=2763](https://web.archive.org/web/20081201065328/http://www.batteryspace.com/index.asp?PageAction=VIEWPROD&ProdID=2763)). Archived from the original (<http://www.batteryspace.com/index.asp?PageAction=VIEWPROD&ProdID=2763>) on 2008-12-01. Retrieved 2012-12-14.
23. "Lithium Batteries" ([https://web.archive.org/web/20110808112807/http://www.globalspec.com/Specifications/Electrical\\_Electronic\\_Components/Batteries/Lithium\\_Batteries](https://web.archive.org/web/20110808112807/http://www.globalspec.com/Specifications/Electrical_Electronic_Components/Batteries/Lithium_Batteries)). Archived from the original ([http://www.globalspec.com/Specifications/Electrical\\_Electronic\\_Components/Batteries/Lithium\\_Batteries](http://www.globalspec.com/Specifications/Electrical_Electronic_Components/Batteries/Lithium_Batteries)) on 2011-08-08. Retrieved 2010-07-02.
24. Justin Lemire-Elmore (2004-04-13). "The Energy Cost of Electric and Human-Powered Bicycles" ([https://web.archive.org/web/20120913095738/http://www.ebikes.ca/sustainability/Ebike\\_Energy.pdf](https://web.archive.org/web/20120913095738/http://www.ebikes.ca/sustainability/Ebike_Energy.pdf)) (PDF). p. 7. Archived from the original ([http://www.ebikes.ca/sustainability/Ebike\\_Energy.pdf](http://www.ebikes.ca/sustainability/Ebike_Energy.pdf)) (PDF) on 2012-09-13. Retrieved 2009-02-26. "Table 3: Input and Output Energy from Batteries"
25. "Lithium Sulfur Rechargeable Battery Data Sheet" ([https://web.archive.org/web/20080828105501/http://www.sionpower.com/pdf/sion\\_product\\_spec.pdf](https://web.archive.org/web/20080828105501/http://www.sionpower.com/pdf/sion_product_spec.pdf)) (PDF). Sion Power, Inc. 2005-09-28. Archived from the original ([http://www.sionpower.com/pdf/sion\\_product\\_spec.pdf](http://www.sionpower.com/pdf/sion_product_spec.pdf)) (PDF) on 2008-08-28.
26. Kolosnitsyn, V.S.; E.V. Karaseva (2008). "Lithium-sulfur batteries: Problems and solutions". *Russian Journal of Electrochemistry*. 44 (5): 506–509. doi:10.1134/s1023193508050029 (<https://doi.org/10.1134%2Fs1023193508050029>). S2CID 97022927 (<https://api.semanticscholar.org/CorpusID:97022927>).
27. "Storage Technology Report, ST6 Flywheel" (<https://web.archive.org/web/20130114062530/http://www.itpower.co.uk/investire/pdfs/flywheelrep.pdf>) (PDF). Archived from the original (<http://www.itpower.co.uk/investire/pdfs/flywheelrep.pdf>) (PDF) on 2013-01-14. Retrieved 2012-12-14.
28. "Next-gen Of Flywheel Energy Storage" (<https://web.archive.org/web/20100710052927/http://www.pddnet.com/article-next-gen-of-flywheel-energy-storage/>). Product Design & Development. Archived from the original (<http://www.pddnet.com/article-next-gen-of-flywheel-energy-storage/>) on 2010-07-10. Retrieved 2009-05-21.
29. "Advanced Materials for Next Generation NiMH Batteries, Ovonic, 2008" (<https://web.archive.org/web/20100104134725/http://ovonic.com/PDFs/ovonic-materials/Ovonic-Fetcenko-2008-Wolsky-Seminar.pdf>) (PDF). Archived from the original (<http://www.ovonic.com/PDFs/ovonic-materials/Ovonic-Fetcenko-2008-Wolsky-Seminar.pdf>) (PDF) on 2010-01-04. Retrieved 2012-12-14.
30. "ZBB Energy Corp" (<https://web.archive.org/web/20071015134212/http://zbbenergy.com/technology.htm>). Archived from the original (<http://www.zbbenergy.com/technology.htm>) on 2007-10-15. "75 to 85 watt-hours per kilogram"
31. High Energy Metal Hydride Battery ([http://www.movitrom.com/files\\_pdf/baterias/saft/NHE\\_en.pdf](http://www.movitrom.com/files_pdf/baterias/saft/NHE_en.pdf)) Archived ([https://web.archive.org/web/20090930120510/http://www.movitrom.com/files\\_pdf/baterias/saft/NHE\\_en.pdf](https://web.archive.org/web/20090930120510/http://www.movitrom.com/files_pdf/baterias/saft/NHE_en.pdf)) 2009-09-30 at the Wayback Machine
32. "Microsoft Word - V-FUEL COMPANY AND TECHNOLOGY SHEET 2008.doc" (<https://web.archive.org/web/20101122114137/http://www.vfuel.com.au/infosheet.pdf>) (PDF). Archived from the original (<http://www.vfuel.com.au/infosheet.pdf>) (PDF) on 2010-11-22. Retrieved 2010-05-07.
33. "Maxwell Technologies: Ultracapacitors - BCAP3000" (<http://maxwell.com/ultracapacitors/products/large-cell/bcap3000.asp>). Maxwell.com. Retrieved 2010-05-07.
34. "Archived copy" (<https://web.archive.org/web/20120722130618/http://www3.fs.cvut.cz/web/fileadmin/documents/12241-BOZEK/publikace/2004/Sup-Cap-Energy-Storage.pdf>) (PDF). Archived from the original (<http://www2.fs.cvut.cz/web/fileadmin/documents/12241-BOZEK/publikace/2004/Sup-Cap-Energy-Storage.pdf>) (PDF) on 2012-07-22. Retrieved 2012-12-14.

35. [1] ([http://www.accel.de/pages/2\\_mj\\_superconducting\\_magnetic\\_energy\\_storage\\_smes.html](http://www.accel.de/pages/2_mj_superconducting_magnetic_energy_storage_smes.html)) Archived ([https://web.archive.org/web/20100216175256/http://www.accel.de/pages/2\\_mj\\_superconducting\\_magnetic\\_energy\\_storage\\_smes.html](https://web.archive.org/web/20100216175256/http://www.accel.de/pages/2_mj_superconducting_magnetic_energy_storage_smes.html)) February 16, 2010, at the Wayback Machine
  36. "Department of Computing" ([https://web.archive.org/web/20061006125946/http://www.doc.ic.ac.uk/~mpj01/ise2grp/energystorage\\_report/node9.html](https://web.archive.org/web/20061006125946/http://www.doc.ic.ac.uk/~mpj01/ise2grp/energystorage_report/node9.html)). Archived from the original ([http://www.doc.ic.ac.uk/~mpj01/ise2grp/energystorage\\_report/node9.html](http://www.doc.ic.ac.uk/~mpj01/ise2grp/energystorage_report/node9.html)) on 2006-10-06. Retrieved 2012-12-14.
  37. "Archived copy" (<https://web.archive.org/web/20110513205201/http://www.askmar.com/Magnets/Promising%20Magnet%20Applications.pdf>) (PDF). Archived from the original (<http://www.askmar.com/Magnets/Promising%20Magnet%20Applications.pdf>) (PDF) on 2011-05-13. Retrieved 2012-12-14.
  38. "Garage Door Springs" (<http://garagedoor.org/residential/torsion-springs.php>). Garagedoor.org. Retrieved 2010-05-07.
- 

Retrieved from "[https://en.wikipedia.org/w/index.php?title=Energy\\_density\\_Extended\\_Reference\\_Table&oldid=1153898740](https://en.wikipedia.org/w/index.php?title=Energy_density_Extended_Reference_Table&oldid=1153898740)"